

Subscriber access provided by ISTANBUL TEKNIK UNIV

Bioactive Terpenoids from Octocorallia, I. Bioactive Diterpenoids: Litophynols A and B from the Mucus of the Soft Coral Litophyton sp.

Tomofumi Miyamoto, Koji Yamada, Nobuya Ikeda, Tetsuya Komori, and Ryuichi Higuchi

J. Nat. Prod., 1994, 57 (9), 1212-1219• DOI: 10.1021/np50111a004 • Publication Date (Web): 01 July 2004

Downloaded from http://pubs.acs.org on April 4, 2009

More About This Article

The permalink http://dx.doi.org/10.1021/np50111a004 provides access to:

- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article



Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

BIOACTIVE TERPENOIDS FROM OCTOCORALLIA, I. BIOACTIVE DITERPENOIDS: LITOPHYNOLS A AND B FROM THE MUCUS OF THE SOFT CORAL *LITOPHYTON* SP.

Tomofumi Miyamoto, Koji Yamada, Nobuya Ikeda, Tetsuya Komori, and Ryuichi Higuchi*

Faculty of Pharmaceutical Sciences, Kyushu University, Maidashi 3-1-1, Higashi-ku, 812 Fukuoka, Japan

ABSTRACT.—Two new eunicellin-type diterpenoids, litophynols A [1] and B [2], and the known diterpenoids litophynins E [3], H [11], and I monoacetate [12], were isolated from the mucus secreted by the soft coral *Litophyton* sp. Their structures have been elucidated on the basis of spectral and single crystal X-ray analyses. These compounds have hemolytic activity.

The litophynins, which are biologically active diterpenoids of the eunicellin class (1), have been isolated from the whole bodies of the soft coral *Litophyton* sp. (Octocorallia) collected from Sukumo Bay in Kochi Prefecture by Ochi *et al.* (2,3). This animal, named "Numeri-Tosaka" in Japanese, secretes much mucus from its surface when stimulated. In this paper, we report the structures of two new eunicellin-type diterpenoids, litophynols A [1] and B [2], isolated from the mucus of the soft coral *Litophyton* sp., which was collected from the Nichinan Coast in Miyazaki Prefecture. The absolute configuration of litophynin E [3] (4) was also determined, and the biological activities of these diterpenoids are discussed.

RESULTS AND DISCUSSION

The Et_2O -soluble part of the EtOH extract obtained from the mucus of *Litophyton* sp. (4.2 kg) was subjected to Sephadex LH-20 and Si gel cc to furnish diterpenoid



fractions, which showed ichthyotoxic and hemolytic activities. These fractions were purified by reversed-phase column chromatography by mplc (RP-8) and hplc ($(8NVC_{18})$) guided by hemolytic activity to give two novel eunicellin-type diterpenoids, litophynols A [1] and B [2], together with three known diterpenoids, litophynin E [3], H [11] (5), and litophynin I monoacetate [12] (6).

A molecular formula of $C_{24}H_{38}O_5$ for litophynol A [1] was determined by hreims data (m/z 406.2697 [M⁺]). The ir spectrum of 1 exhibited absorptions due to hydroxy (3610 cm⁻¹) and ester (1730 cm⁻¹) functionalities. The ¹H- and ¹³C-nmr spectra of 1 suggested the presence of one primary methyl, two secondary methyls, one tertiary methyl, six methylenes, four methines, four oxygen-bearing methines, one oxygen-bearing quaternary carbon, two exocyclic methylenes, and one ester (Table 1).

The fdms spectrum of 1 showed the presence of a butyloxy functionality [m/z 319 $[M^+ - 87], m/z 71 (C_3H_7CO^+)]$. Acetylation of 1 with Ac₂O/pyridine gave the diacetate [4]. These data indicated that 1 is a tricyclic diterpenoid possessing two hydroxy groups and one butyloxy group. Since diacetate 4 afforded crystals suitable for X-ray analysis, the determination of the relative stereostructure of 4 was conducted by this means. The results revealed the relative stereostructure of 4 as shown in Figure 1.

The absolute configuration of **1** was determined by application of the cd exciton chirality method (7). Perbenzoylation of **1** afforded a dibenzoate [**5**]. The cd spectrum of **5** showed a positive chirality ($[\theta]_{238} + 76600$, $[\theta]_{224} - 53000$) due to the C-6 and C-8 benzoates. These data suggested that the absolute configurations of C-6 and C-8 were S and R, respectively (see Figure 2).

The molecular formula of litophynol B [2] was determined to be $C_{24}H_{40}O_6$, by hreims data (*m*/z 424.2827 [M⁺]). Compound 2 exhibited similar spectral features to those of 1 except for signals due to C-6–C-7–C-8. The ¹³C-¹H long-range COSY nmr spectrum revealed that the quaternary methyl signal δ_H 1.25 (3H, s, Me-16) had correlations with two oxygen-bearing methine signals δ_C 79.7 (d, C-8) and δ_C 77.3 (d, C-6). Furthermore, the oxygen-bearing methine signal δ_H 3.55 (1H, d, *J*=9.2 Hz, H-8) correlated with another oxymethine signal δ_H 3.88 (1H, dd, *J*=5.9 and 8.9 Hz, H-9) in the ¹H-¹H COSY spectrum. These data suggested that litophynol B [2] had a contiguous triol system in its 9-membered ring.

The relative stereochemistry of **2** was deduced by comparison of ¹H-nmr spectral data for sclerophytin C [**6**] (8), which was isolated from another soft coral *Sclerophytum capitalis*. This comparison indicated that the relative stereochemistries at C-6, -7, and -8 of **2** were the same as those of **6**. The absolute configuration of **2** was determined in the same manner as for **1**. Thus, the cd spectrum of the dibenzoate derivative [**7**] of **2** showed a positive chirality ($[\theta]_{237}$ +33000, $[\theta]_{221}$ -42000) due to the C-6 and C-8 benzoates, which suggested that the absolute configurations of C-6 and C-8 in **2** were *S* and *S*, respectively (see Figure 3).

The absolute configuration of litophynin E [3], which was isolated by Ochi *et al.*, has not yet been determined. We have investigated its absolute configuration at this time. The (+)-(R)-MTPA (8R) and (-)-(S)-MTPA (8S) esters of 3 were prepared, and a modified Mosher method (9) was applied to these MTPA esters. The positive and negative $\Delta\delta$ values were found on the right and left sides of the MTPA planes, respectively, indicating an S configuration for C-6. Thus, the absolute configuration of litophynin E was determined (see Figure 4).

The absolute configurations of these compounds were further confirmed as follows. Ozonolysis of 2 and 3 gave monooxo derivatives 9 and 10, respectively (Scheme 1). Their structures were suggested by the ir absorptions due to the ketone group at C-11 (1715 cm⁻¹; 9, 1720 cm⁻¹; 10). The cd spectra of 9 and 10 showed similar negative Cotton

				Com	punod			
Position		1		2		3		6
	δ _c	δ _H (J, Hz)	$\boldsymbol{\delta}_{\mathrm{c}}$	δ _H (J, Hz)	$\delta_{\rm c}$	δ _H (J, Hz)	δ _c	δ _H (J, Hz)
1	44.0 (d)		45.2 (d)		45.5 (d)	2.19 dd (4.0, 7.0)	45.0 (d)	
2	91.4 (d)	3.75 s	(p) 6.16	3.60 s	92.1 (d)	3.58 s	91.4 (d)	3.58 s
3 5	84.6 (s)		86.1 (s)		86.5 (s)		86.2 (s)	
4	35.2 (t)		35.2 (t)		36.2 (t)		34.5 (t)	
5	28.7 (t)		29.6 (t)		30.5 (t)		29.5 (t)	
	(P) 8:99	4.72 dd (4.0, 11.0)	77.3 (d)	4.61 d (6.0)	80.2 (d)	4.58 d (6.0)	77.0 (d)	4.61 d (6.4)
7	152.1 (s)		79.8 (s)		76.9 (s)		79.6 (s)	
	77.4 (d)	4.16 s	(p) 16.7 (d)	3.55 d (9.0)	45.9 (t)	1.81 dd (4.0, 15.0)	79.5 (d)	3.54 d (8.8)
						1.91 dd (11.0, 15.0)		
	83.7 (d)	4.16 d (10.5)	81.2 (d)	3.88 dd (6.0, 9.0)	78.3 (d)	4.14 ddd	81.1 (d)	3.86 dd (6.3, 8.8)
						(4.0, 7.0, 11.0)		
10	48.0 (d)	2.84 dd (8.0, 10.5)	52.9 (d)	3.29 t (6.5)	53.7 (d)	2.97 t (7.0)	52.5 (d)	3.25 dd (6.4, 6.5)
11	146.0 (s)		148.8 (s)		147.6 (s)		148.6 (s)	
12	31.6 (t)		31.7 (t)		31.5 (t)		31.6 (t)	
13	25.3 (t)		24.9 (t)		24.6 (t)		24.8 (t)	
14	44.4 (d)	1.28 br t	43.9 (d)		44.0 (d)		43.7 (d)	
15	22.2 (q)	1.61 s	23.1 (q)	1.41 s	23.2 (q)	1.40 s	23.0 (q)	1.40 s
16	118.2 (t)	5.22 s, 5.52 s	17.6 (q)	1.25 s	22.7 (q)	1.16 s	17.7 (g)	1.23 s
17	111.6 (t)	4.65 s, 4.81 s	110.3 (t)	4.79 s, 4.89 s	109.4 (t)	4.65 s, 4.69 s	109.9 (t)	4.78 s, 4.87 s
18	27.5 (d)	1.87 m	29.1 (d)		29.1 (d)		29.0 (d)	
19	15.5 (q)	0.75 d (6.5)	16.1 (q)	0.80 d (6.5)	15.7 (q)	0.79 d (6.5)	16.2 (q)	0.80 d (6.5)
20	21.9 (q)	0.97 d (7.0)	22.0 (q)	0.97 d (7.0)	22.0 (q)	0.97 d (7.0)	21.9 (q)	0.96 d (6.5)
1'	172.6 (s)		172.4 (s)		172.3 (s)		169.5 (s)	
2'	37.4 (t)		37.4 (t)		37.4 (t)		22.7 (q)	2.01 s
3'	18.5 (t)		18.4 (t)		18.4 (t)			
4'	13.6 (q)	0.92 t (7.5)	13.7 (q)	0.98 t (7.5)	13.7 (q)	0.99 t (7.5)		

TABLE 1. ¹³C-Nmr and ¹H-Nmr Data for Littophynols A [1] and B [2], Littophynin E [3], and Sclerophytin C [6].



FIGURE 1. ORTEP view of 4.

effect curves at 294 nm ($\Delta \epsilon$ ext -0.7; **9**) and at 297 nm ($\Delta \epsilon$ ext -0.5; **10**), and application of the octant rule (10) to their cd spectra supported the absolute configuration determined as described above.

We investigated the constituents of the animal bodies, from which the mucus was removed. Five eunicellin-type diterpenoids (1-3, 11, and 12) were also obtained in low yield compared with the mucus. Litophynols A [1] and B [2], litophynins E [3] and H [11], and I monoacetate [12] were positive in a hemolytic reaction test, and crude diterpenoid fractions exhibited ichthyotoxicity (IC₁₀₀ 20 ppm). This suggests that this soft coral holds eunicellin-type diterpenoids in its mucus for the purpose of defense against predators.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES .- Mps were determined on a Yanaco micro-melting point



FIGURE 2. The dihedral angle between C-6 and C-8 in benzoate was examined by nOe experiments, J values, and MM2 (13) calculations.



FIGURE 3. The dihedral angle between C-6 and C-8 in benzoate was examined by nOe experiments, J values, and MM2 (13) calculations.



FIGURE 4. $\Delta\delta$ Values ($\Delta\delta = \delta_s - \delta_R$ in Hz at 270 MHz) obtained for MTPA esters $\mathbf{8}_R / \mathbf{8}_s$.

apparatus and are uncorrected. Spectra were recorded on the following instruments: specific optical rotations, Jasco DIP-370 digital polarimeter; circular dichroism, Jasco J-600 spectrophotometer; ir, Jasco IR-810 spectrophotometer; eims/hreims, JEOL JMS-DX-300/JMA-3500 data system, accelerating potential of 3 kV, ionizing potential of 30 eV, sample temperature of 200–250°; nmr, JEOL FX-270 (270 MHz) spectrometer in CDCl₃. X-ray crystallographic measurements were made on a Rigaku RASA-5R automatic single-crystal X-ray structure analysis system. Normal and reversed-phase tlc were performed with Merck Si gel 60 F₂₅₄ and RP-8 F₂₅₄, respectively. Cc was carried out with Pharmacia Sephadex LH-20, Merck Si gel 60 (0.063–0.200 μ m), and Fuji Davison Si gel BW-300. Hplc and mplc separations were performed on a Jasco BIP 1 hplc pump and a RID-300 RI detector. The columns were Merck LiChroprep RP-8 (40–63 μ m) and Waters radial pak 8NVC₁₈ (4 μ m) (RCM 8×10).

COLLECTION AND TAXONOMY.—The soft coral is probably an undescribed species of the genus *Litophyton* (phylum Cnidaria, class Anthozoa, subclass Octocorallia, order Alcyonaceae, family Nephtheidae), and was collected from the rocky coast of Nango-cho, Miyazaki Prefecture, Japan, at 3 m deep, in November 1989. It has a short sterile stalk and bush-like capitulum. Unfortunately, the base and the lower part of the stalk were missing. The capitulum was covered with non-retractile monomorphic zooids, 0.3 mm in diameter, 1.2 mm in height including tentacles. Sclerites of the stalk cortex are warty spindles, 0.03×0.17 mm. The stalk interior contained a few spindles, 0.02×0.11 mm. The capitulum was free from sclerites. The color of the colony is creamy white when stored in alcohol. The present specimen is closely related to *Litophyton viridis* (May, 1898) in the absence of the sclerite in the capitulum, but is different in the size of sclerite at the stalk. A voucher specimen is deposited in the Wakayama Prefecture Museum of Natural History (Catalog No. 1994-INV-0004).

EXTRACTION AND ISOLATION.—The mucus (4.2 kg) was separated from the animal (7.9 kg, wet wt) and extracted with EtOH (14 liters). The EtOH layer was evaporated *in vacuo* and the residue partitioned between H_2O (1 liter) and Et_2O (1.8 liter). The Et_2O layer was evaporated *in vacuo* to give 20.6 g of extract. Part of this (11.5 g) was chromatographed on a Sephadex LH-20 column with CHCl₃-MeOH (1:1) as eluent to yield three fractions: fractions 1 (941.9 mg), 2 (10.43 g), and 3 (146.0 mg). Fraction 2 was



chromatographed on a Si gel column. Elution was performed with *n*-hexane-EtOAc (4:1 \rightarrow EtOAc) to yield two main fractions which exhibited hemolytic activity: fraction 4 (352.6 mg) and fraction 5 (275.8 mg). Fraction 4 was chromatographed on LiChroprep RP-8 with 85% aqueous MeOH, followed by column chromatography on Si gel (BW-300) with *n*-hexane-EtOAc (3:2) to give pure litophynin E [**3**] (33.3 mg). Fraction 5 was chromatographed on LiChroprep RP-8 with 85% aqueous MeOH to give two hemolytic fractions: fraction 6 (95.2 mg) and fraction 7 (35.6 mg). Fraction 6 was subjected to hplc (8NVC₁₈) with 70% aqueous MeOH to give litophynol B [**2**] (36.2 mg), litophynin H [**11**] (7.8 mg), and litophynin I monoacetate [**12**] (5.0 mg). Fraction 7 was chromatographed on Si gel (BW-300) with *n*-hexane-EtOAc (1:2) to give litophynol A [**1**] (26.8 mg).

HEMOLYTIC ACTIVITY.—Performed with a 2% erythrocyte suspension: fresh rabbit blood (1 ml) was washed three times with 0.1 M phosphate-buffered saline of pH 6.8 (PBS), and resuspended in 50 ml of 0.85% saline. Sample solutions were subjected to tlc with a suitable developing solvent, and the 2% erythrocyte suspension was sprayed on the plate. After 15 min, hemolysins were observed as pale red spots.

ICHTHYOTOXICITY.—Ichthyotoxicity assays were conducted using the mosquito fish, *Oryzias latipes*. Fractions 2, 4, and 5 were assayed at 20 ppm by dissolving the appropriate amount in 1 ml of EtOH. Control tests were carried out with each test run. The toxicity was evaluated as times for lethality to occur. Fractions 4 and 5 killed fish within 24 h at 20 ppm.

Litophynol A [1].—Amorphous solid; mp 131–132°; $[\alpha]^{28}D + 19.2°(c=1.1, CHCl_3)$; ir (CHCl₃) ν max 3610, 3100, 1730, 1655, 910 cm⁻¹; fdms *m/z* 406 {M⁺}, 389 [M–17], 319 {M–87}; hreims *m/z*, found {M⁺} 406.2697 (C₂₄H₃₈O₃ requires 406.2718); ¹H nmr and ¹³C nmr, see Table 1.

Acetylation of 1.—Litophynol A [1] (13.9 mg) was dissolved in pyridine (1.5 ml) and added to (Ac₂O) (1.5 ml). The reaction mixture was stood for 24 h at room temperature, and was then poured into excess H₂O (50 ml). The aqueous layer was extracted with CHCl₃ (20, 10, 10 ml). The CHCl₃ layer was evaporated *in vacuo* to yield the crude extract, which was treated by chromatography on Si gel using *n*-hexane-EtOAc (3:1) as eluent to yield a pure diacetate [4] (10.0 mg) as colorless plates; $[\alpha]^{27}D - 22.4^{\circ}$ (*c*=0.9, CHCl₃); fdms *m/z* 490 [M]⁺, 431 [M-59], 403 [M-87]; ¹H nmr (270 MHz, CDCl₃) δ 0.76 (3H, d, *J*=6.9 Hz, CH₃-19), 0.91 (3H, t, *J*=7.6 Hz, H-4'), 0.97 (3H, d, *J*=7.3 Hz, CH₃-20), 1.65 (3H, s, H-15), 1.97 (3H, s, COCH₃), 2.06 (3H, s, COCH₃), 2.95 (1H, dd, *J*=11.1 and 7.8 Hz, H-10), 3.82 (1H, s, H-2), 4.17 (1H, d, *j*=10.7 Hz, H-9), 4.72 (1H, s, H-17), 4.86 (1H, t, *J*=1.7 Hz, H-17), 5.19 (1H, s, H-8), 5.49 (1H, s, H-16), 5.66 (1H, s, H-16), 5.79 (1H, dd, *J*=11.0 and 4.1 Hz, H-6).

X-RAY CRYSTAL STRUCTURE ANALYSIS OF 4.—Crystal data: $C_{28}H_{42}O_7$; mol wt=490.64, orthorhombic, space group P2₁2₁2₁, No. 19, a=18.80 (2), b=23.04 (2), c=6.54 (1) Å (from 15 orientation reflections, 9.2°<20<11.1°), V=2832 (7) Å3, Z=4, Dc=1.151g/cm³, m(MoK α radiation, λ =0.71069 Å). Intensity data were recorded on a Rigaku AFC 5R diffractometer [MoK α radiation, ω =2 θ scans, θ max=55.1°, scanwidth (1.42+0.30 tan θ °)]. The intensities of three standard reflections re-measured every 150 reflections during data collection to monitor crystal stability, indicated that significant deterioration occurred (overall intensity loss = 0.5%). From a total of 3210 measurements, those 1200 reflections with I>3.00 σ (I) were retained for the analysis. Lorentz-polarization corrections were applied.

The crystal structure was solved by direct methods (SIR88)(12). The non-hydrogen atoms were refined anisotropically. The final cycle of full-matrix least-squares refinement was based on 1200 observed reflections and 388 variable parameters and converged with unweighted and weighted agreement factors of: R=0.067, Rw=0.076, GOF=2.73.

Crystallographic calculations were performed on a Micro-VAX 3200 by use of the TEXSAN Structure Analysis Software. In the least-squares iterations, $\Sigma \omega(|Fo|-|Fc|)2$, $\omega = 4Fo^2/\sigma^2(Fo^2)$ was minimized.

Benzoylation of 1.—Litophynol A [1] (2.5 mg) was dissolved in pyridine (5 µl) and added to benzoyl chloride (8 µl). The reaction mixture was stood for 24 h at room temperature, and was then poured into excess H₂O (50 ml). The aqueous layer was extracted with CHCl₃ (15, 10, 10 ml). The CHCl₃ layer was evaporated *in vacuo* to yield a crude extract, which was subjected to hplc [radial pak 8NVC₁₈ (4 µm)] using 90% aqueous MeOH as eluent to yield a pure dibenzoate [**5**] (0.6 mg) which exhibited: fdms *m*/z 614 [M]⁺, 493 [M-121]; ¹H nmr (270 MHz, CDCl₃) δ 0.78 (3H, d, *J*=6.6 Hz), 0.95 (3H, t, *J*=7.3 Hz), 0.98 (3H, d, *J*=6.6 Hz), 1.77 (3H, s), 3.08 (1H, dd, *J*=10.9 and 7.6 Hz), 3.90 (1H, s), 4.36 (1H, d, *J*=9.9 Hz), 4.83 (1H, s), 4.93 (1H, s), 5.57 (1H, s), 5.65 (1H, s), 5.85 (1H, s), 6.33 (1H, dd, *J*=11.2 and 4.0 Hz), 7.0–8.0 (10H, m); cd (*c*=6.0×10⁻⁶, MeOH), $\Delta \varepsilon$ = +23.2 (λ ext = 238 nm), $\Delta \varepsilon$ = -16.1 (λ ext = 224 nm).

Litophynol B [2].—Colorless oil; $[\alpha]^{2^8}D - 17.6^\circ$ (c=3.1, CHCl₃); ir ν max (CHCl₃) 3590, 3090, 1735, 1650, 905 cm⁻¹; fdms *m*/z 424 [M]⁺, 406 [M-18], 337 [M-87]; hreims *m*/z found [M]⁺ 424.2827 ($C_{24}H_{40}O_6$ requires 424.2823); ¹H nmr and ¹³C nmr, see Table 1.

Benzoylation of 2.—Litophynol B [2] (3.1 mg) was benzoylated as described above and purified with hplc [radial pak 8NVC₁₈ (4 μm)] using 90% aqueous MeOH as eluent to yield a pure dibenzoate [7] (2.5 mg) which exhibited: fdms m/z 632 [M]⁺, 511 [M-121]; ¹H nmr (270 MHz, CDCl₃), 0.78 (3H, d, J=6.9 Hz), 0.99 (3H, d, J=6.9 Hz), 1.07 (3H, t, J=7.6 Hz), 1.27 (3H, s), 1.43 (3H, s), 3.51 (1H, t, J=7.0 Hz), 3.90 (1H, s), 3.72 (1H, s), 4.34 (3H, m), 5.61 (1H, d, J=9.6 Hz), 6.03 (1H, dd, J=2.4 and 5.4 Hz), 7.4-8.1 (10H, m); cd (c=2.5 × 10⁻⁵, MeOH), Δε = +9.9 (λ ext = 237 nm), Δε = -12.8 (λ ext = 221 nm).

Litophynin E [3].—Colorless oil; $[\alpha]^{2^7}D + 12.6^{\circ}(c=3.1, \text{CHCl}_3)$; ir $\nu \max (\text{CHCl}_3)$ 3600, 3100, 1735, 1650, 900 cm⁻¹; fdms *m/z* 408 [M]⁺, 390 [M-18], 321 [M-87]; hreims *m/z* found [M]⁺ 408.2869 (C₂₄H₄₀O₅ requires 408.2874); ¹H nmr and ¹³C nmr, see Table 1.

Preparation of (R)- and (S)-MTPA esters of litophynin E [3].—Litophynin E [3] (8.0 mg, 19.6 μ mol), (+)-MTPA acid (11.4 mg, 48.7 mmol), dicyclohexylcarbodiimide (10.7 mg, 51.9 μ mol), and 4-(dimethylamino)pyridine (10.3 mg, 84.3 μ mol) were dissolved in CH₂Cl₂(5 ml). The reaction mixture was stirred in an atmosphere of N₂ for 113 h at room temperature, and the residue obtained after evaporation of the solvent was treated by chromatography on Si gel using *n*-hexane-Me₂CO (95:5) as eluent to yield the pure (*R*)-MTPA [8₈] (5.1 mg, 64%) and (S)-MTPA esters [8₅] (35%).

(R)-*MTPA* ester [**8**_g].—¹H nmr (270 MHz, CDCl₃) δ 0.78 (3H, d, J=6.9 Hz), 0.98 (3H, d, J=6.9 Hz), 1.00 (3H, t, J=7.3 Hz), 1.13 (3H, s), 1.28 (1H, m), 1.39 (3H, s), 1.57 (2H, m), 1.72 (2H, m), 1.73 (1H, m), 1.87 (2H, m), 2.17 (1H, m), 2.18 (1H, m), 2.32 (2H, m), 2.69 (1H, dd, J=14.5 and 7.6 Hz), 3.00 (1H, t, J=7.3 Hz), 3.58 (3H, s, MTPA), 3.66 (1H, s), 4.15 (1H, m), 4.66 (1H, s), 4.69 (1H, s), 5.86 (1H, d, J=5.0 Hz), 7.41 (3H, m), 7.54 (2H, m).

(S)-*MTPA* ester [8₅].—¹H nmr (270 MHz, CDCl₃) δ 0.79 (3H, d, J=6.8 Hz), 0.98 (3H, d, J=6.8 Hz), 0.99 (3H, t, J=7.3 Hz), 1.16 (3H, s), 1.28 (1H, m), 1.40 (3H, s), 1.48 (2H, m), 1.68 (2H, m), 1.76 (1H, m), 1.90 (2H, m), 2.15 (1H, m), 2.21 (1H, m), 2.32 (2H, m), 2.68 (1H, ddd, J=15.3, 8.5, and 3.1 Hz), 3.03 (1H, t, J=7.1 Hz), 3.52 (3H, s, MTPA), 3.66 (1H, s), 4.15 (1H, m), 4.66 (1H, s), 4.71 (1H, s), 5.84 (1H, d, J=5.0 Hz), 7.40 (3H, m), 7.55 (2H, m).

Ozonolysis of litophynol A [2] and litophynin E [3].—Ozone was bubbled through a solution of 2 (5.8 mg) in 1 ml of MeOH at -78° until the solution colored pale blue. The residue obtained after evaporation of the solvent was treated with dimethyl sulfide (10 µl) for 12 h, and then poured into excess H₂O (50 ml). The aqueous layer was extracted with CHCl₃ (10, 5, 5 ml). The CHCl₃ layer was evaporated *in vacuo* to yield a crude extract, which was treated by chromatography on Si gel using *n*-hexane-Me₂CO (1:1) as eluent to yield the pure oxo-derivative of **2** [9] (1.5 mg, 26%) and the oxo-derivative of **3** [10] (1.9 mg, 53%).

 $\begin{array}{l} Oxo-derivative 9. \\ -Ir \ \nu \ max (CHCl_3) 3600, 1735, 1720 \ cm^{-1}; cd \ (c=1.6\times10^{-4}, MeOH), \Delta \varepsilon = -0.5 \\ (\lambda \ ext = 297 \ nm); fabms (positive ion, m/z) 409 \ [M-17, base peak]; ^{1}H \ nmr (270 \ MHz, CDCl_3) \delta 0.90 \ (3H, d, J=6.6 \ Hz), 0.97 \ (3H, t, J=7.3 \ Hz), 1.05 \ (3H, d, J=6.6 \ Hz), 1.27 \ (3H, s), 1.48 \ (3H, s), 2.56 \ (1H, dt, J=4.6 \ and 17.2 \ Hz), 2.70 \ (1H, ddd, J=3.3, 8.9, and 8.9 \ Hz), 3.11 \ (1H, dd, J=4.4 \ and 7.9 \ Hz), 3.17 \ (1H, d, J=4.6 \ Hz), 3.51 \ (1H, dd, J=4.6 \ and 9.6 \ Hz), 3.81 \ (1H, d, J=3.3 \ Hz), 4.15 \ (1H, dd, J=4.6 \ and 9.6 \ Hz), 4.57 \ (1H, br \ d, J=7.6 \ Hz); ^{13}C \ nmr \ (67.5 \ MHz, CDCl_3) \delta 44.6 \ (d, C-1), 90.6 \ (d, C-2), 86.0 \ (s, C-3), 34.0 \ (t, C-4), 29.3 \ (t, C-5), 77.4 \ (d, C-6), 78.6 \ (s, C-7), 75.8 \ (d, C-8), 80.2 \ (d, C-9), 57.2 \ (d, C-10), 213.2 \ (s, C-11), 38.0 \ (t, C-12), 22.1 \ (t, C-13), 42.3 \ (d, C-14), 23.2 \ (q, C-15), 17.9 \ (q, C-16), 28.8 \ (d, C-18), 16.7 \ (q, C-19), 21.9 \ (q, C-20), 172.1 \ (s, C-1'), 37.3 \ (t, C-2'), 18.3 \ (t, C-3'), 13.6 \ (q, C-4'). \end{array}$

Oxo-derivative **10**.—Ir ν max (CHCl₃) 3600, 1735, 1715 cm⁻¹; cd ($c=1.9 \times 10^{-4}$, MeOH), $\Delta \varepsilon = -0.7$ (λ ext = 294 nm); fabms (positive ion, m/z) 393 [M-17, base peak]; ¹H nmr (270 MHz, CDCl₃) δ 0.89 (3H, d, J=6.9 Hz), 0.97 (3H, t, J=7.3 Hz), 1.05 (3H, d, J=6.9 Hz), 1.16 (3H, s), 1.50 (3H, s), 2.70 (1H, ddd, J=3.0, 7.9, and 7.9 Hz), 2.77 (1H, dd, J=4.6 and 7.6 Hz), 3.76 (1H, d, J=3.0 Hz), 4.51 (1H, br d, J=8.3 Hz), 4.59 (1H, dt, J=4.2 and 11.6 Hz); ¹³C nmr (67.5 MHz, CDCl₃) δ 45.7 (d, C-1), 90.7 (d, C-2), 86.0 (s, C-3), 34.1 (t, C-4), 30.3 (t, C-5), 78.1 (d, C-6), 76.3 (s, C-7), 46.1 (t, C-8), 77.2 (d, C-9), 57.9 (d, C-10), 210.6 (s, C-11), 37.8 (t, C-12), 22.6 (t, C-13), 42.7 (d, C-14), 23.4 (q, C-15), 22.6 (q, C-16), 28.9 (d, C-18), 17.0 (q, C-19), 22.0 (q, C-20), 172.3 (s, C-1'), 37.3 (t, C-2'), 18.4 (t, C-3'), 13.7 (q, C-4').

Litophynin H **[11**].—Amorphous solid; $[\alpha]^{27}D + 10.6^{\circ}(c=0.5, CHCl_3)$; ir $\nu \max (CHCl_3) 3600, 1730, 1650, 920 cm^{-1}$; fdms *m/z* 406 [M]⁺ 389 [M-17], 319 [M-87]; hreims *m/z* found [M]⁺ 406.2740 (C₂₄H₃₈O, requires 406.2717); ¹H nmr (270 MHz, CDCl_3) 0.76 (3H, d, J=6.6 Hz), 0.92 (3H, t, J=7.4 Hz), 0.99 (3H, d, J=6.6 Hz), 1.64 (3H, s), 3.07 (1H, dd, J=10.6 and 8.3 Hz), 3.79 (1H, s), 4.41 (2H, m), 4.54 (1H, dd, J=10.4 and 4.1 Hz), 4.86 (1H, d, J=1.3 Hz), 5.11 (1H, s), 5.45 (1H, s); ¹³C nmr (67.5 MHz, CDCl₃) & 43.6 (d, C-1), 90.2 (d, C-2), 84.7 (s, C-3), 35.6 (t, C-4), 28.5 (t, C-5), 72.2 (d, C-6), 151.0 (s, C-7), 38.5 (t, C-8), 81.9 (d, C-9), 45.8 (d, C-10), 146.6 (s, C-11), 71.7 (d, C-12), 31.1 (t, C-13), 35.9 (d, C-14), 22.1 (q, C-15), 116.6 (t, C-3'), 13.6 (q, C-4').

Litophynin I monoacetate [12].—Colorless oil; $[\alpha]^{2^7}D + 40.1^{\circ}(c=0.3, CHCl_3)$; ir $\nu \max (CHCl_3) 3600$, 1740, 1655, 910 cm⁻¹; fdms m/z 466 [M]⁺, 449 [M-17], 406 [M-60], 379 [M-87]; hreims m/z found [M]⁺ 466.2962 (C₂₆H₄₂O₇ requires 466.2928); ¹H nmr (270 MHz, CDCl_3) δ 0.82 (3H, d, J=6.9 Hz), 0.96 (3H, d, J=6.9 Hz), 0.98 (3H, t, J=7.4 Hz), 1.17 (3H, s), 1.45 (3H, s), 2.07 (3H, s), 2.97 (1H, t, J=6.6 Hz), 3.69 (1H, d, J=1.7 Hz), 4.39 (1H, m), 4.58 (1H, d, J=5.9 Hz), 4.97 (1H, s), 5.11 (1H, s), 5.15 (1H, s), 5.45 (1H, dd, J=5.0 and 2.6 Hz); ¹³C nmr (67.5 MHz, CDCl_3) δ 44.3 (d, C-1), 90.5 (d, C-2), 86.4 (s, C-3), 35.3 (t, C-4), 30.5 (t, C-5), 79.4 (d, C-6), 77.2 (s, C-7), 45.9 (t, C-8), 78.9 (d, C-9), 51.5 (d, C-10), 143.0 (s, C-11), 72.6 (d, C-12), 28.9 (t, C-13), 37.0 (d, C-14), 23.3 (q, C-15), 22.7 (q, C-16), 115.3 (t, C-17), 28.8 (d, C-18), 16.0 (q, C-19), 21.8 (q, C-20), 172.2 (s, C-1'), 37.4 (t, C-2'), 18.4 (t, C-3'), 13.7 (q, C-4'), 170.3 (s, Ac), 21.5 (q, Ac).

ACKNOWLEDGMENTS

We thank Dr. Yukio Imahara, Wakayama Prefectural Museum of Natural History, for identification of the soft coral. This work was supported in part by the Grant-in-Aid for Scientific Research (No. 3214 and 06453217) from the Ministry of Education, Science, and Culture, Japan, which is gratefully acknowledged.

LITERATURE CITED

- 1. D.B. Stierle, B. Carte, D.J. Faulkner, B. Tagel, and J. Clardy, J. Am. Chem. Soc., 102, 5088 (1980).
- 2. M. Ochi, K. Futatsugi, H. Kotsuki, M. Ishii, and K. Shibata, Chem. Lett., 2207 (1987).
- 3. M. Ochi, K. Futatsugi, Y. Kume, H. Kotsuki, K. Asao, and K. Shibata, Chem. Lett., 1661 (1988).
- 4. M. Ochi, K. Yamada, K. Futatsugi, H. Kotsuki, and K. Shibata, Chem. Lett., 2183 (1990).
- 5. M. Ochi, K. Yamada, K. Futatsugi, and H. Kotsuki, Heterocycles, 32, 29 (1991).
- 6. M. Ochi, K. Yamada, K. Kataoka, H. Kotsuki, and K. Shibata, Chem. Lett., 155 (1992).
- N. Harada and K. Nakanishi, "Circular Dichroic Spectroscopy-Exciton Coupling in Organic Stereochemistry," University Science Books, Mill Valley, CA, and Oxford University Press, Oxford, UK, 1983.
- 8. M. Alam, P. Sharma, A.S. Zektzer, G.E. Martin, X. Ji, and D. van der Helm, *J. Org. Chem.*, **54**, 1896 (1989).
- 9. I. Ohtani, T. Kusumi, Y. Kashman, and H. Kakisawa, J. Am. Chem. Soc., 113, 4092 (1991).
- W. Moffit, R.B. Woodward, A. Moscowitz, W. Klyne, and C. Djerassi, J. Am. Chem. Soc., 83, 4013 (1961).
- 11. M.C. Burla, M. Camalli, G. Cascarano, C. Giacovazzo, G. Polidori, R. Spagna, and D. Viterbo, J. Appl. Cryst., 22, 389 (1989).
- C.K. Johnson, ORTEP-II, a FORTRAN Thermal-Ellipsoid Plot Program, Report ORNL-5138, Oak Ridge National Laboratory, Oak Ridge, Tennessee, 1976.
- 13. N.L. Allinger, J. Am. Chem. Soc., 99, 8127 (1977).

Received 8 February 1994